

BEER

Current quality assessment:

appearance: color, foam, clarity
taste: sweetness, sourness, saltiness, bitterness
flavor, aroma

Factors affecting chemical composition:

water quality, malt, hop, yeasts
recipe and timing of the brewing process

Motivation:

the relationship of the current quality properties with chemical composition is not fully understood

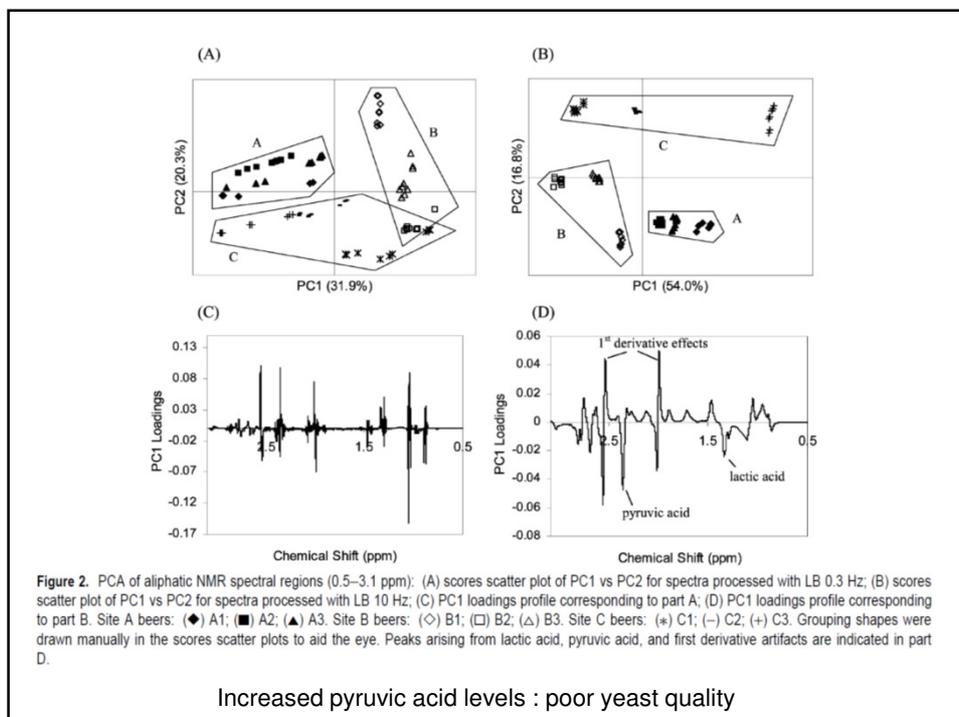
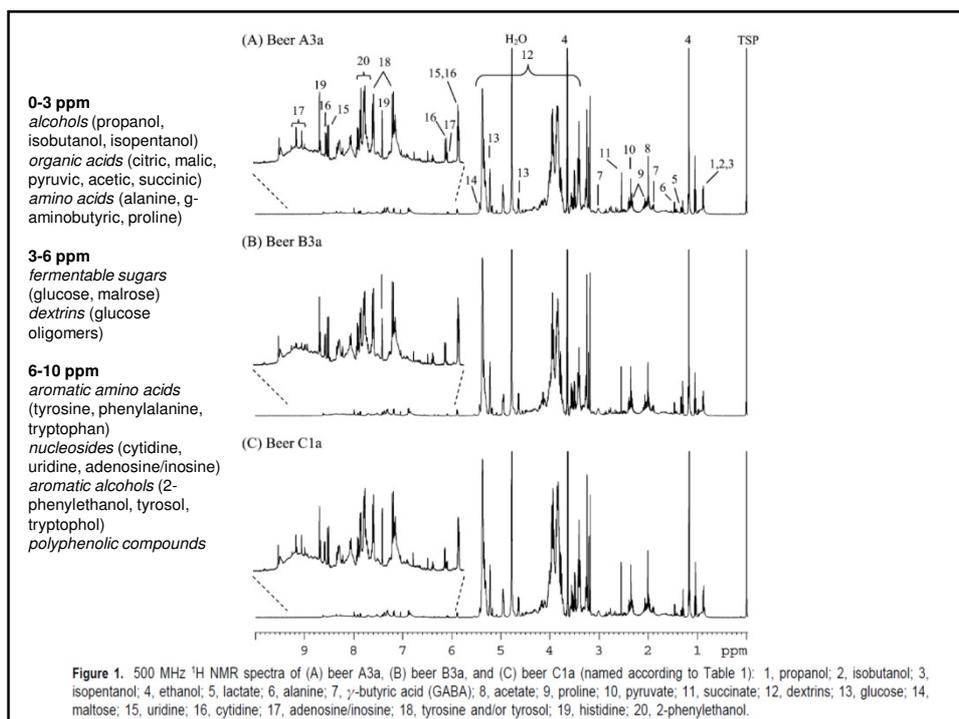
700 *J. Agric. Food Chem.* 2006, 54, 700–706

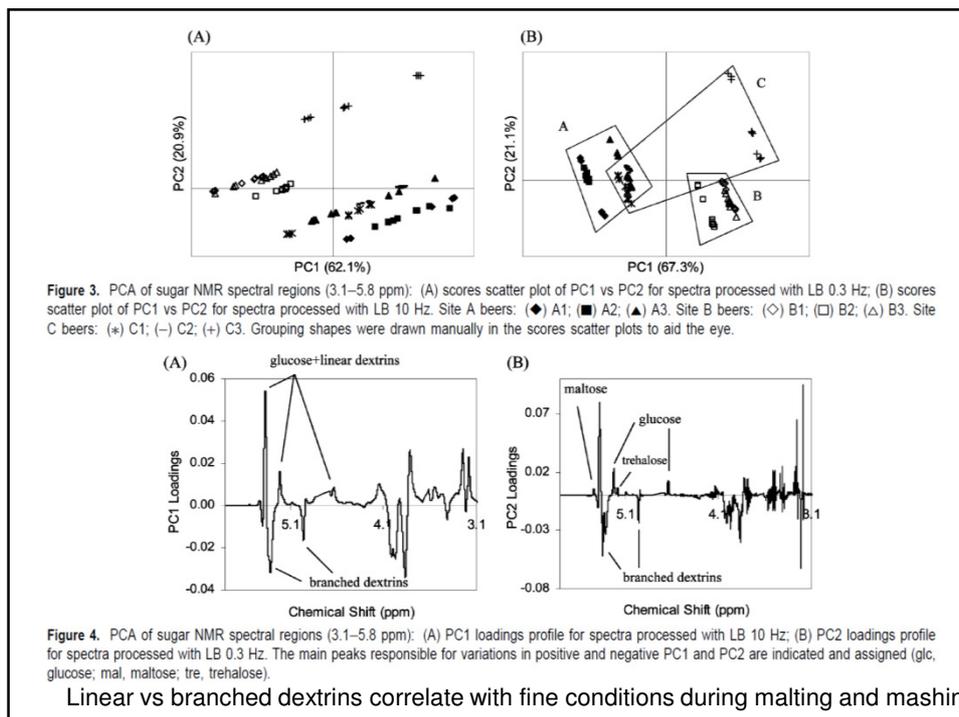
JOURNAL OF
**AGRICULTURAL AND
FOOD CHEMISTRY**

**Composition of Beer by ^1H NMR Spectroscopy: Effects of
Brewing Site and Date of Production**

CLÁUDIA ALMEIDA,[†] IOLA F. DUARTE,[†] ANTÓNIO BARROS,[‡] JOÃO RODRIGUES,[†]
MANFRED SPRUAL,[§] AND ANA M. GIL^{*,†}

CICECO and QOPNAA, Department of Chemistry, Campus Universitário de Santiago,
University of Aveiro, 3810-193 Aveiro, Portugal and Bruker Biospin GmbH, Silberstreifen,
D76287 Rheinstetten, Germany





ORANGE JUICE

580

J. Agric. Food Chem. **2001**, *49*, 580–588

Discrimination between Orange Juice and Pulp Wash by ^1H Nuclear Magnetic Resonance Spectroscopy: Identification of Marker Compounds

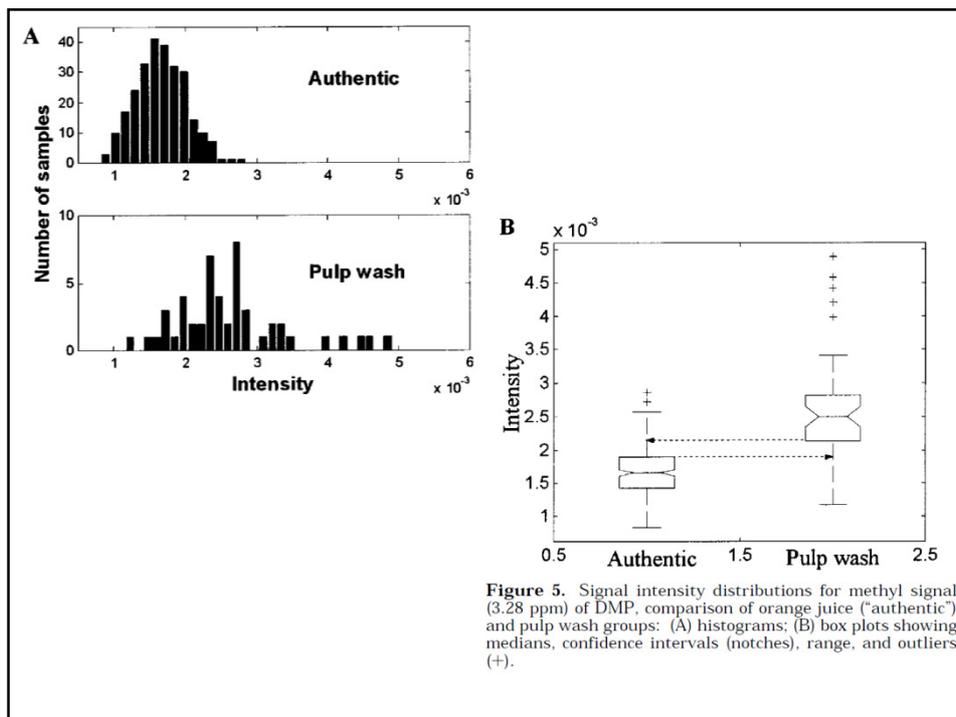
Gwénaëlle Le Gall, Max Puaud, and Ian J. Colquhoun*

Institute of Food Research, Norwich Research Park, Colney, Norwich NR4 7UA, United Kingdom

According to U.S. Food and Drug Administration (FDA) investigations, some companies are known to have made millions of dollars selling fraudulent orange juice

Adulteration may be done by the addition of water, sugars, pulp wash, or other citrus fruit juices

Pulp wash is a second extract obtained by washing the separated pulp with water after the first pressing. Its chemical composition is similar to orange juice but paler, more bitter, and is regarded as lower quality



BLACK TEA

White, green, oolong, black tea differ in the fermentation process:

green: unfermented

white: lightly f.

oolong: partially

black: fermented

All derive from *Camellia sinensis*

Black tea is more oxidized, has stronger flavor, and contains more caffeine

Drinking black tea is associated with reduced cardiovascular risk

During manufacture, enzyme-catalyzed oxidation and partial polymerization of flavonols occur. As a result, theaflavins (TFs) and thearubigins characteristic of the black tea taste and color are produced.

Flavonoids constitute 10-12% of dry leaf weight.

The taste differs according to differences in growing environment.

Characterization of Tea Cultivated at Four Different Altitudes Using ^1H NMR Analysis Coupled with Multivariate Statistics

Akiko Ohno,^{*,†} Kitaro Oka,[‡] Chiseko Sakuma,[§] Haruhiro Okuda,[†] and Kiyoshi Fukuhara^{*,†}

[†]Division of Organic Chemistry, National Institute of Health Sciences, Setagaya-ku, Tokyo 158-8501, Japan

[‡]Department of Clinical Pharmacology and [§]Central Analytical Laboratory, School of Pharmacy, Tokyo University of Pharmacy and Life Science, 1432-1 Horinouchi, Hachioji, Tokyo 192-0392, Japan

Sri-Lanka tea-planting regions:

RAN (>1900m), UDA (1000-1500m), MEDA (600-1200m), YATA (<600m)

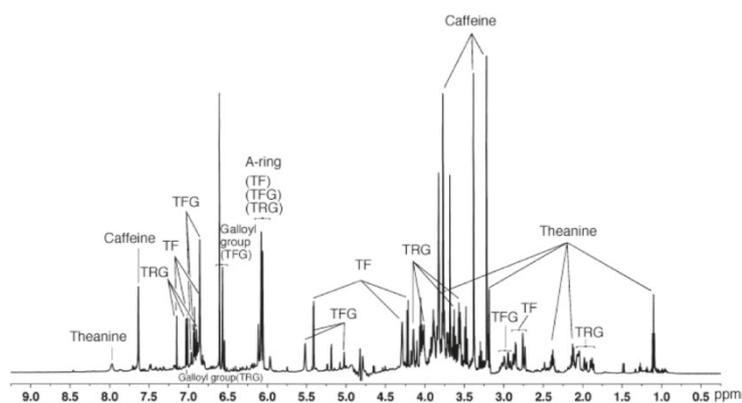
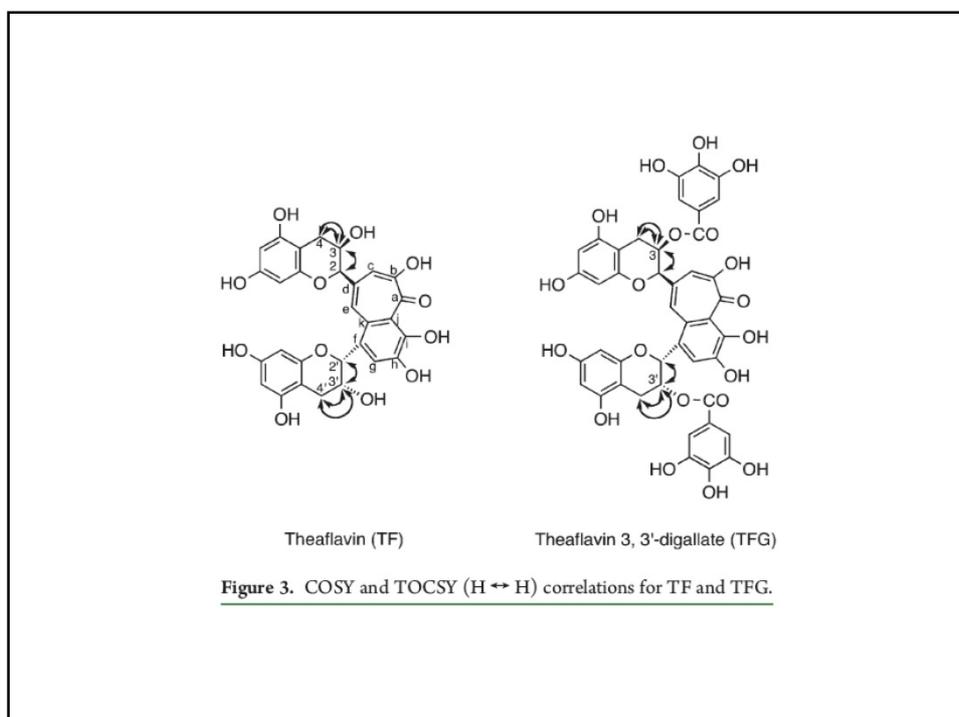
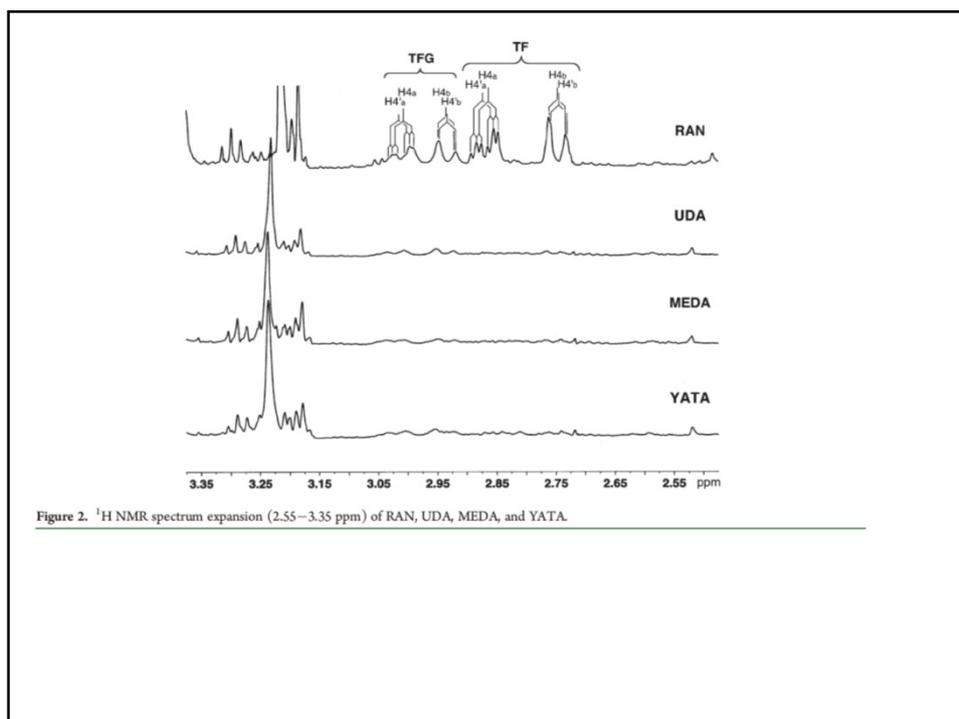
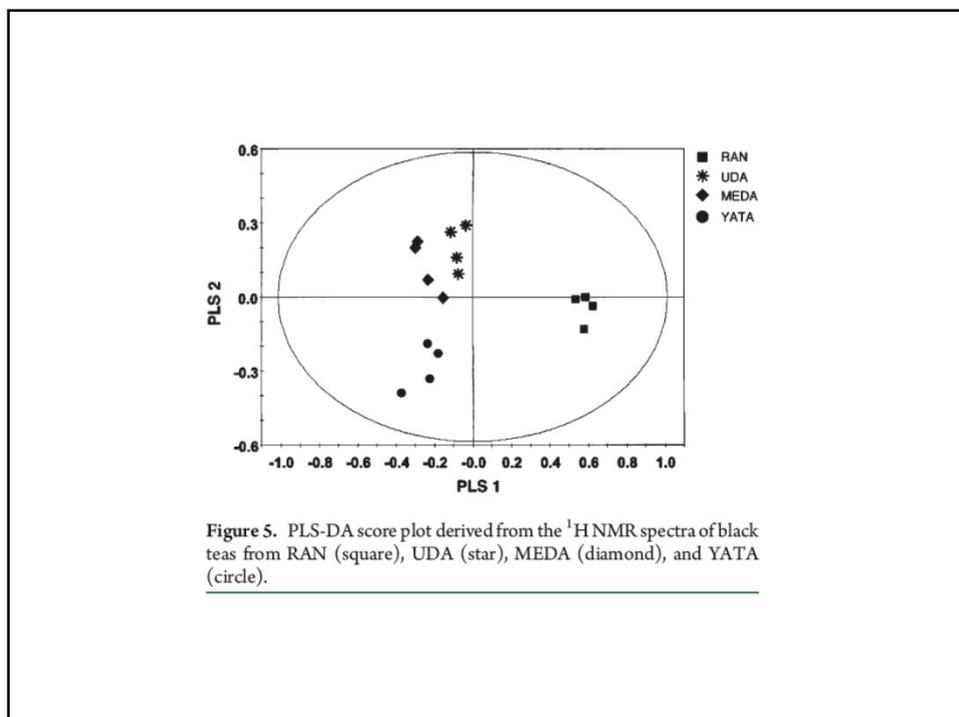
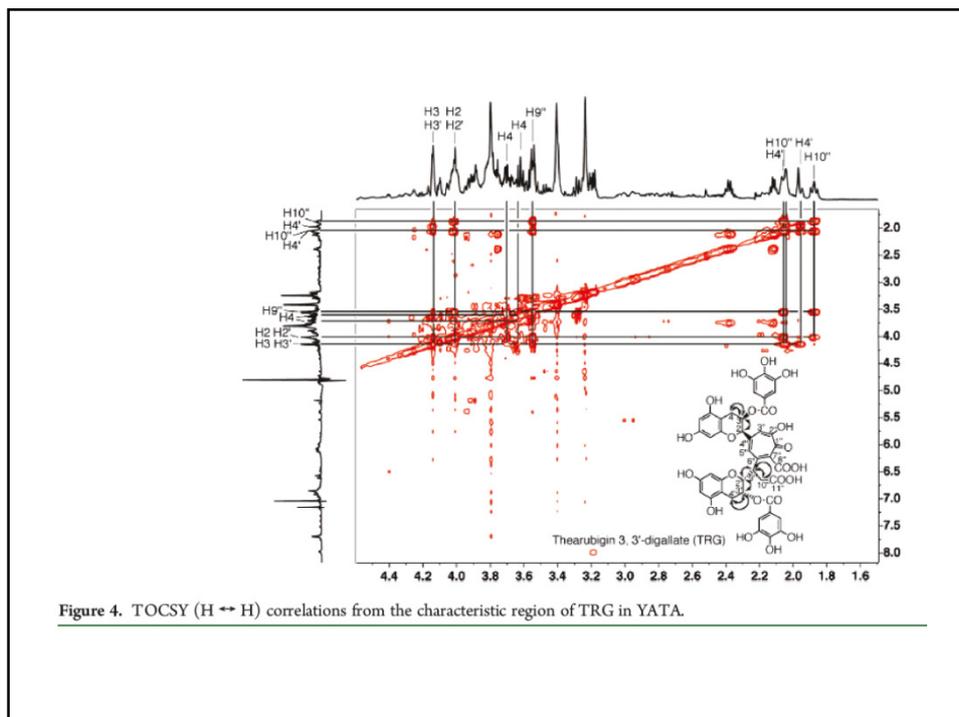
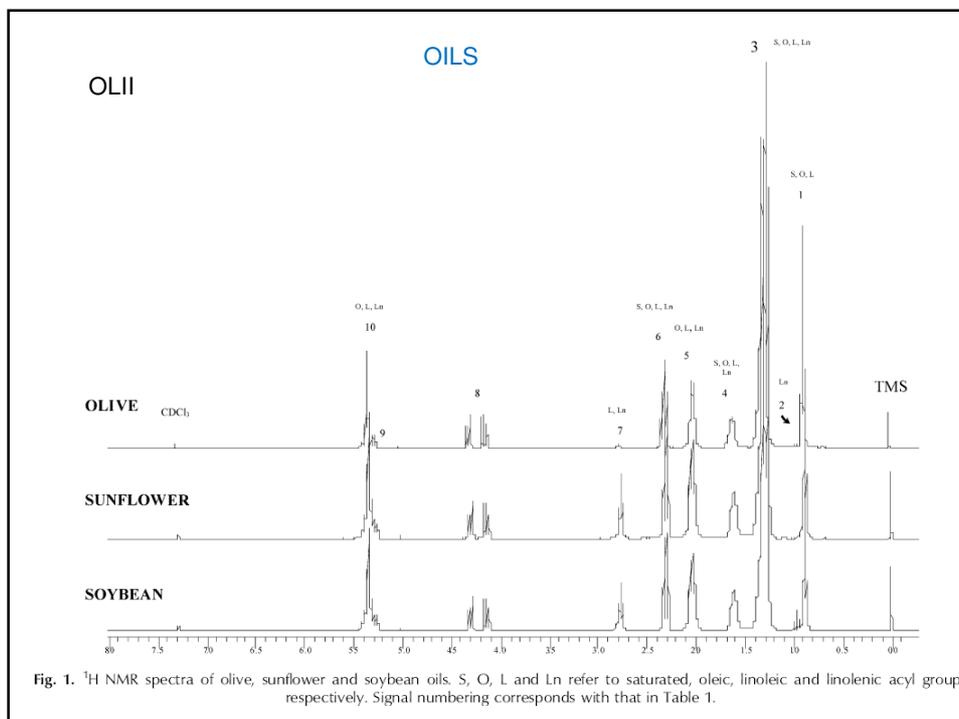
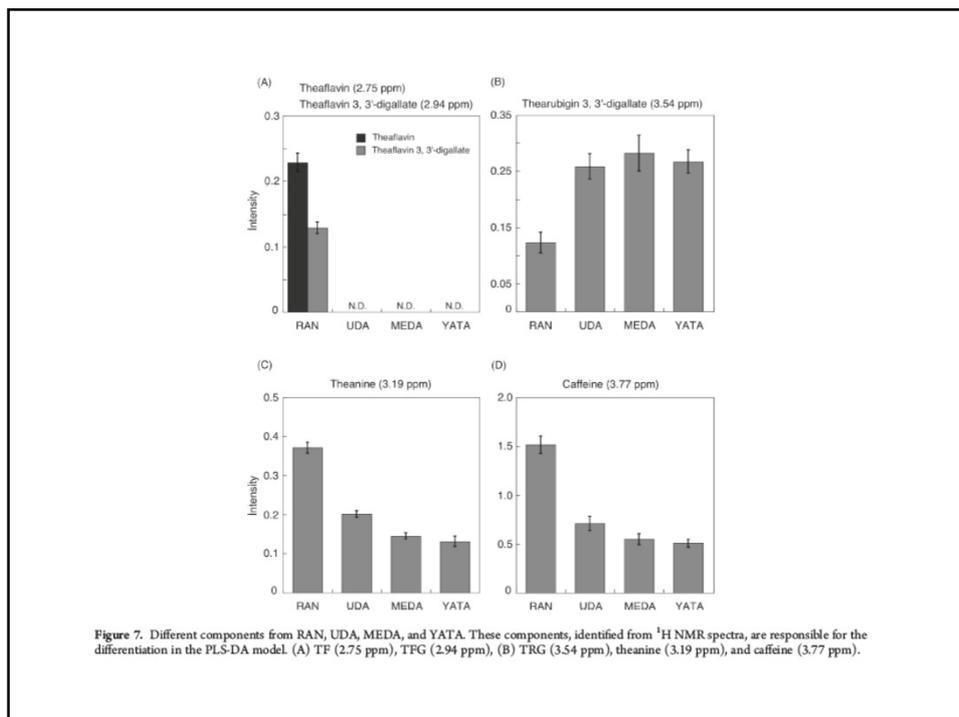


Figure 1. Representative ^1H NMR spectrum of black tea from RAN.







OLII

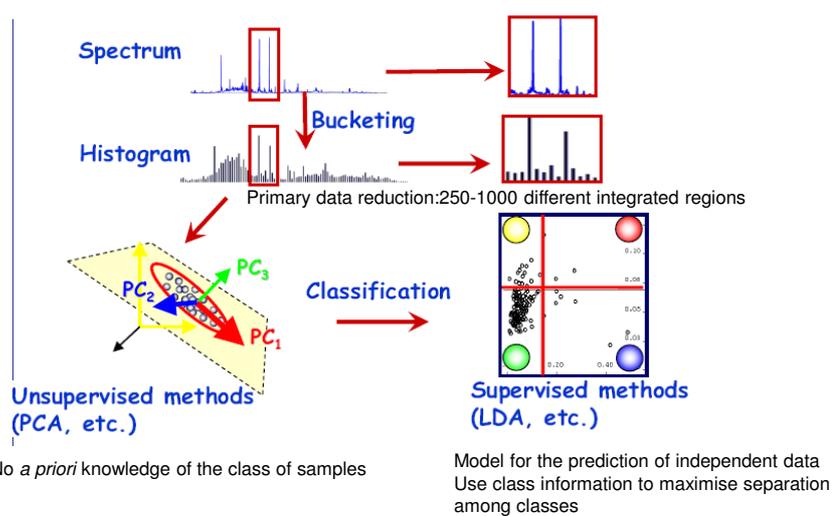
Table 1. Chemical shift assignments of the ^1H NMR signals of the main components of edible oils and fats

Signal	Chemical shift (ppm)	Functional group	Intensity ^a	Authors
1	0.90–0.80	–CH ₃ (acyl group)	m	Segre and Mannina (1997)
1.a	0.823	saturated and oleic (or ω -9)		
1.b	0.839	linoleic (or ω -6)		
2	1.00–0.90	–CH ₃ (acyl group)	v	Segre and Mannina (1997)
2.a	0.925	linolenic (or ω -3)		
3	1.40–1.15	–(CH ₂) _n – (acyl group)	l	Segre and Mannina (1997)
3.a	1.194	saturated		
3.b	1.230	oleic		
3.c	1.280	linoleic and linolenic		
4	1.70–1.50	–OCO–CH ₂ –CH ₂ – (acyl group)	m	Segre and Mannina (1997)
4.a	1.553	saturated		
4.b	1.557	oleic		
4.c	1.567	linoleic and linolenic		
5	2.10–1.90	–CH ₂ –CH=CH– (acyl groups)	m	Segre and Mannina (1997)
5.a	1.948	oleic		
5.b	1.996	linoleic		
5.c	1.994 and 2.030	linolenic		
6	2.35–2.20	–OCO–CH ₂ – (acyl group)	m	Segre and Mannina (1997)
6.a	2.219	saturated		
6.b	2.226	oleic		
6.c	2.238	linoleic and linolenic		
—	2.38 ^b	–OCO–CH ₂ –CH ₂ – (docosahexaenoic acyl groups)	v	Aursand <i>et al.</i> (1993)
7	2.80–2.70	=HC–CH ₂ –CH= (acyl groups)	v	Segre and Mannina (1997)
7.a	2.718	linoleic		
7.b	2.754	linolenic		
8	4.32–4.10	–CH ₂ OCOR (glyceryl group)	m	Segre and Mannina (1997)
9	5.26–5.20	>CHOCOR (glyceryl group)	s	Segre and Mannina (1997)
10	5.40–5.26	–CH=CH– (acyl group)	m	Segre and Mannina (1997)

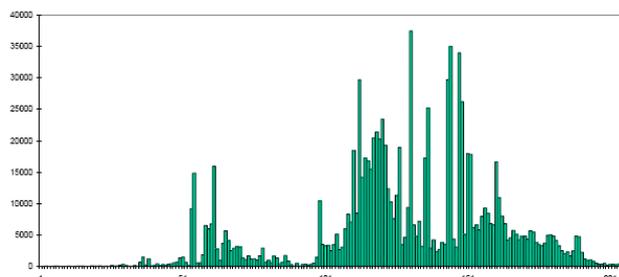
^a l, large; m, medium; s, small; v, variable.

^b Signal only present in fish oils.

NMR-based metabolomics: the concept

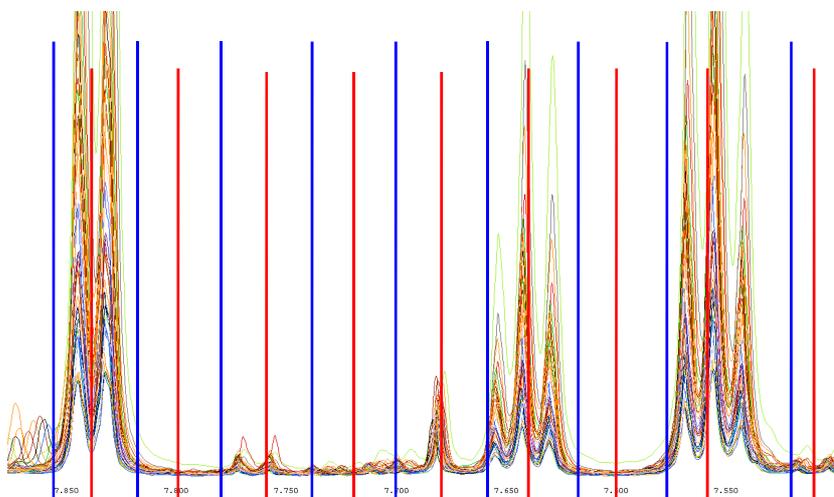


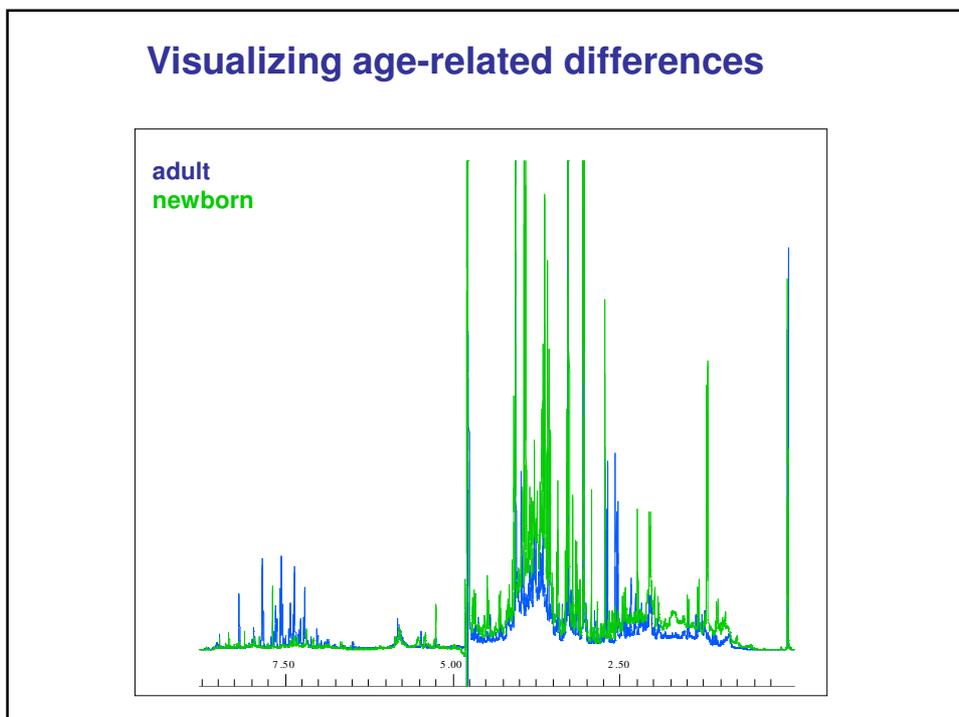
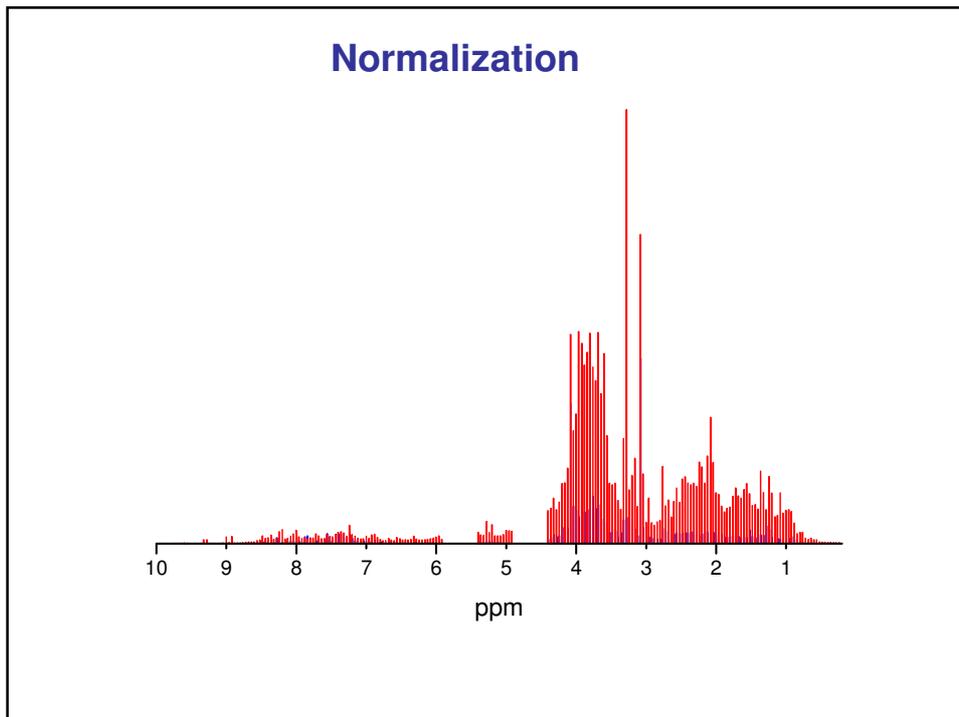
Data pre-processing (NMR)



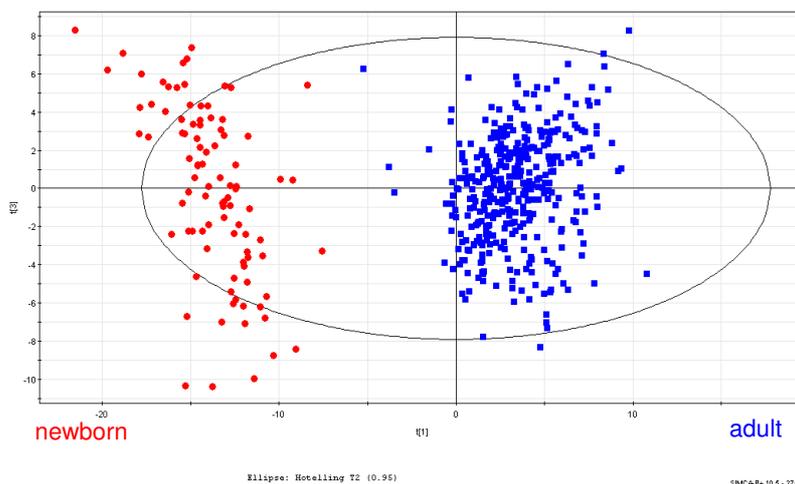
- Discretise x-axis into n equal sized bins, height = area under intensity (reduces impact of small variations in chemical shift e.g. due to pH)
- Normalise bars for constant total area (removes effect of differences in concentration across samples)
- Remove insignificant regions (e.g. water and urea resonances in urine spectra)

Fixed vs variable bucketing





PCA newborns vs adults



Data pre-treatment (general for metabolomics)

Different data preprocessing steps are applied in order to generate 'clean' data in the form of normalized peak areas that reflect the (intracellular) metabolite concentrations. These clean data can be used as the input for data analysis. However, it is important to use an appropriate data pretreatment method before starting data analysis.

Besides induced biological variation, other factors are also present in metabolomics data:

1. **Differences in orders of magnitude between measured metabolite concentrations;** for example, the average concentration of a signal molecule is much lower than the average concentration of a highly abundant compound like ATP. However, from a biological point of view, metabolites present in high concentrations are not necessarily more important than those present at low concentrations.
2. **Differences in the fold changes in metabolite concentration due to the induced variation;** the concentrations of metabolites in the central metabolism are generally relatively constant, while the concentrations of metabolites that are present in pathways of the secondary metabolism usually show much larger differences in concentration depending on the environmental conditions.
3. **Some metabolites show large fluctuations in concentration under identical experimental conditions.** This is called uninduced biological variation.

Besides these biological factors, other effects present in the data set are:

4. **Technical variation;** this originates from, for instance, sampling, sample work-up and analytical errors.
5. **Heteroscedasticity;** for data analysis, it is often assumed that the total uninduced variation resulting from biology, sampling, and analytical measurements is symmetric around zero with equal standard deviations. However, this assumption is generally not true. For instance, the standard deviation due to uninduced biological variation depends on the average value of the measurement. This is called heteroscedasticity, and it results in the introduction of additional structure in the data. Heteroscedasticity occurs in uninduced biological variation as well as in technical variation.

The variation in the data resulting from a metabolomics experiment is the sum of the induced variation and the total uninduced variation. The total uninduced variation is all the variation originating from uninduced biological variation, sampling, sample work-up, and analytical variation. **Data pretreatment focuses on the biologically relevant information by emphasizing different aspects in the clean data.**

... data pre-treatment (general for metabolomics)

Class I: Centering

Centering converts all the concentrations to fluctuations around zero instead of around the mean of the metabolite concentrations. Hereby, it adjusts for differences in the offset between high and low abundant metabolites. It is therefore used to focus on the fluctuating part of the data, and leaves only the relevant variation (being the variation between the samples) for analysis. Centering is applied in combination with all the methods described below.

Class	Method	Formula	Unit	Goal	Advantages	Disadvantages
I	Centering	$\hat{x}_{ij} = x_{ij} - \bar{x}_i$	0	Focus on the differences and not the similarities in the data	Remove the offset from the data	When data is heteroscedastic, the effect of this pretreatment method is not always sufficient
II	Autoscaling	$\hat{x}_{ij} = \frac{x_{ij} - \bar{x}_i}{s_i}$	(-)	Compare metabolites based on correlations	All metabolites become equally important	Inflation of the measurement errors
	Range scaling	$\hat{x}_{ij} = \frac{x_{ij} - \bar{x}_i}{(x_{i_{max}} - x_{i_{min}})}$	(-)	Compare metabolites relative to the biological response range	All metabolites become equally important. Scaling is related to biology	Inflation of the measurement errors and sensitive to outliers
	Pareto scaling	$\hat{x}_{ij} = \frac{x_{ij} - \bar{x}_i}{\sqrt{s_i}}$	0	Reduce the relative importance of large values, but keep data structure partially intact	Stays closer to the original measurement than autoscaling	Sensitive to large fold changes
	Vast scaling	$\hat{x}_{ij} = \frac{(x_{ij} - \bar{x}_i) \cdot s_i}{s_i}$	(-)	Focus on the metabolites that show small fluctuations	Aims for robustness, can use prior group knowledge	Not suited for large induced variation without group structure
	Level scaling	$\hat{x}_{ij} = \frac{x_{ij} - \bar{x}_i}{\bar{x}_i}$	(-)	Focus on relative response	Suited for identification of e.g. biomarkers	Inflation of the measurement errors
III	Log transformation	$\hat{x}_{ij} = \log(x_{ij})$ $\hat{x}_{ij} = x_{ij} - \bar{x}_i$	Log 0	Correct for heteroscedasticity, pseudo scaling. Make multiplicative models additive	Reduce heteroscedasticity, multiplicative effects become additive	Difficulties with values with large relative standard deviation and zero
	Power transformation	$\hat{x}_{ij} = \sqrt[1/2]{x_{ij}}$ $\hat{x}_{ij} = x_{ij} - \bar{x}_i$	$\sqrt{0}$	Correct for heteroscedasticity, pseudo scaling	Reduce heteroscedasticity, no problems with small values	Choice for square root is arbitrary.

van den Berg et al. BMC Genomics 2006 7:142 doi:10.1186/1471-2164-7-142

... data pre-treatment (general for metabolomics)

Class II: Scaling

Scaling methods are data pretreatment approaches that divide each variable by a factor, the scaling factor, which is different for each variable. They aim to adjust for the differences in fold differences between the different metabolites by converting the data into differences in concentration relative to the scaling factor.

There are two subclasses within scaling. The first class uses a measure of the data dispersion (such as, the standard deviation) as a scaling factor, while the second class uses a size measure (for instance, the mean).

Scaling based on data dispersion

Scaling methods tested that use a dispersion measure for scaling were autoscaling, pareto scaling, range scaling, and vast scaling (Table 1). Autoscaling, also called unit or **unit variance scaling**, is commonly applied and uses the standard deviation as the scaling factor. After autoscaling, all metabolites have a standard deviation of one and therefore the data is analyzed on the basis of correlations instead of covariances, as is the case with centering.

Pareto scaling is very similar to autoscaling. However, instead of the standard deviation, the square root of the standard deviation is used as the scaling factor. Now, large fold changes are decreased more than small fold changes, thus the large fold changes are less dominant compared to clean data.

Scaling based on average value

Level scaling falls in the second subclass of scaling methods, which use a size measure instead of a spread measure for the scaling. Level scaling converts the changes in metabolite concentrations into changes relative to the average concentration of the metabolite by using the mean concentration as the scaling factor. The resulting values are changes in percentages compared to the mean concentration. As a more robust alternative, the median could be used. Level scaling can be used when large relative changes are of specific biological interest, for example, when stress responses are studied or when aiming to identify relatively abundant biomarkers.

... data pre-treatment (general for metabolomics)

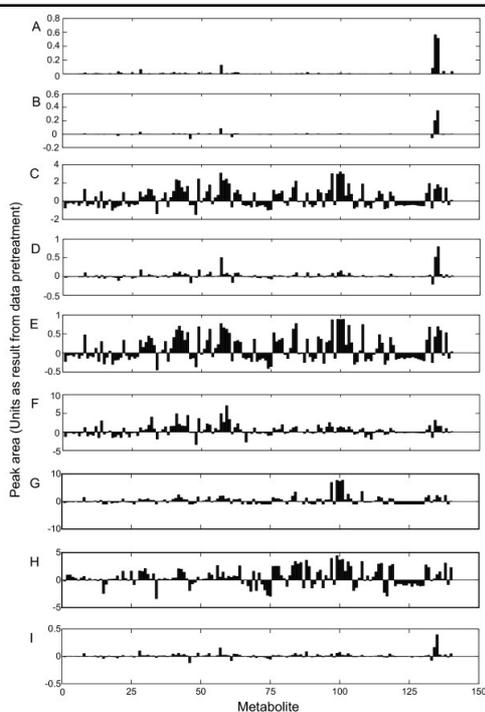
Class III: Transformations

Transformations are nonlinear conversions of the data like, for instance, the log transformation and the power transformation (Table 1). Transformations are generally applied to correct for heteroscedasticity, to convert multiplicative relations into additive relations, and to make skewed distributions (more) symmetric. In biology, relations between variables are not necessarily additive but can also be multiplicative. A transformation is then necessary to identify such a relation with linear techniques.

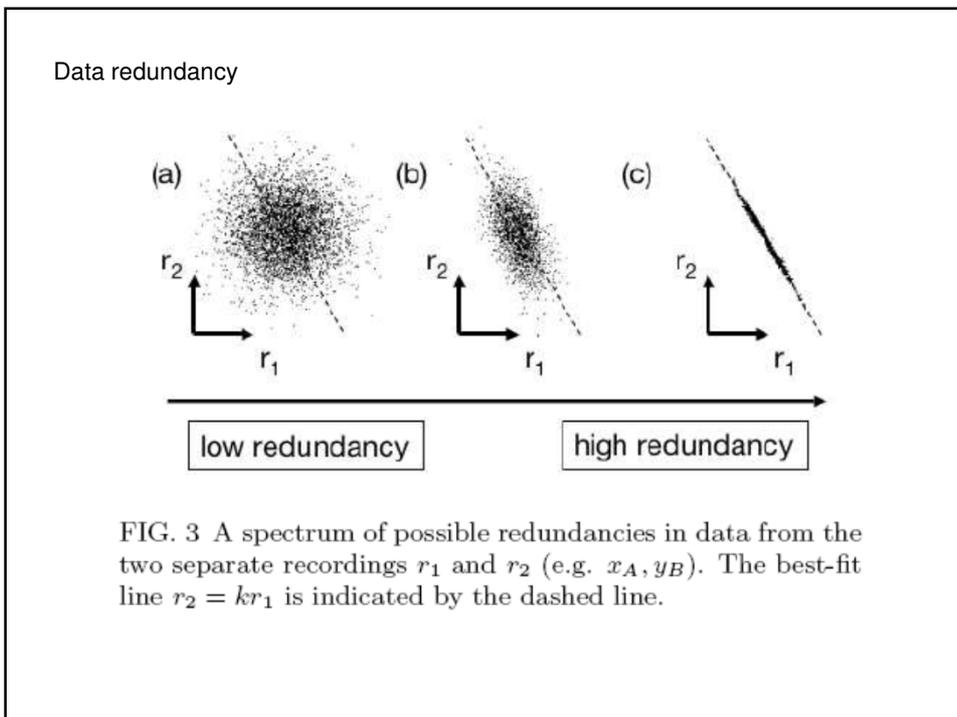
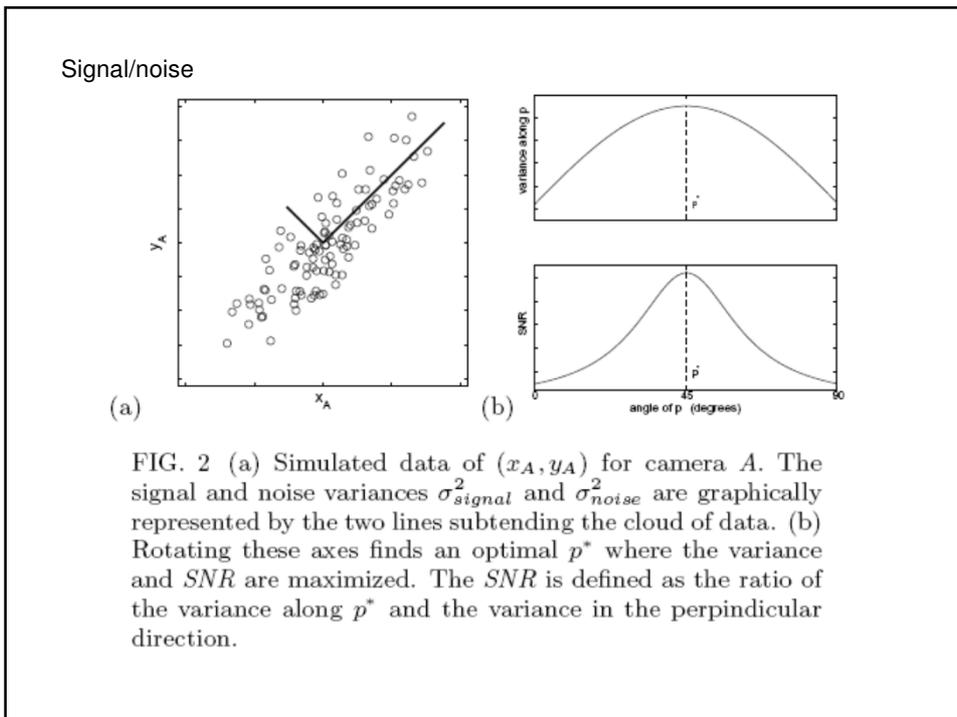
Since the log transformation and the power transformation reduce large values in the data set relatively more than the small values, the transformations have a pseudo scaling effect as differences between large and small values in the data are reduced. However, the pseudo scaling effect is not determined by the multiplication with a scaling factor as for a 'real' scaling effect, but by the effect that these transformations have on the original values. This pseudo scaling effect is therefore rarely sufficient to fully adjust for magnitude differences. Hence, it can be useful to apply a scaling method after the transformation. However, it is not clear how the transformation and a scaling method influence each other with regard to the complex metabolomics data.

A transformation that is often used is the **log transformation** (Table 1). A log transformation perfectly removes heteroscedasticity if the relative standard deviation is constant. However, this is rarely the case in real life situations. A drawback of the log transformation is that it is unable to deal with the value zero. Furthermore, its effect on values with a large relative analytical standard deviation is problematic, usually the metabolites with a relatively low concentration, as these deviations are emphasized. These problems occur because the log transformation approaches minus infinity when the value to be transformed approaches zero.

A transformation that does not show these problems and also has positive effects on heteroscedasticity is the power transformation (Table 1). The power transformation shows a similar transformation pattern as the log transformation. Hence, the power transformation can be used to obtain results similar as after the log transformation without the near zero artifacts, although the power transformation is not able to make multiplicative effects additive.



Effect of data pretreatment on the original data. Original data of experiment G2 (A), and the data after centering (B), autoscaling (C), pareto scaling (D), range scaling (E), vast scaling (F), level scaling (G), log transformation (H), and power transformation (I).



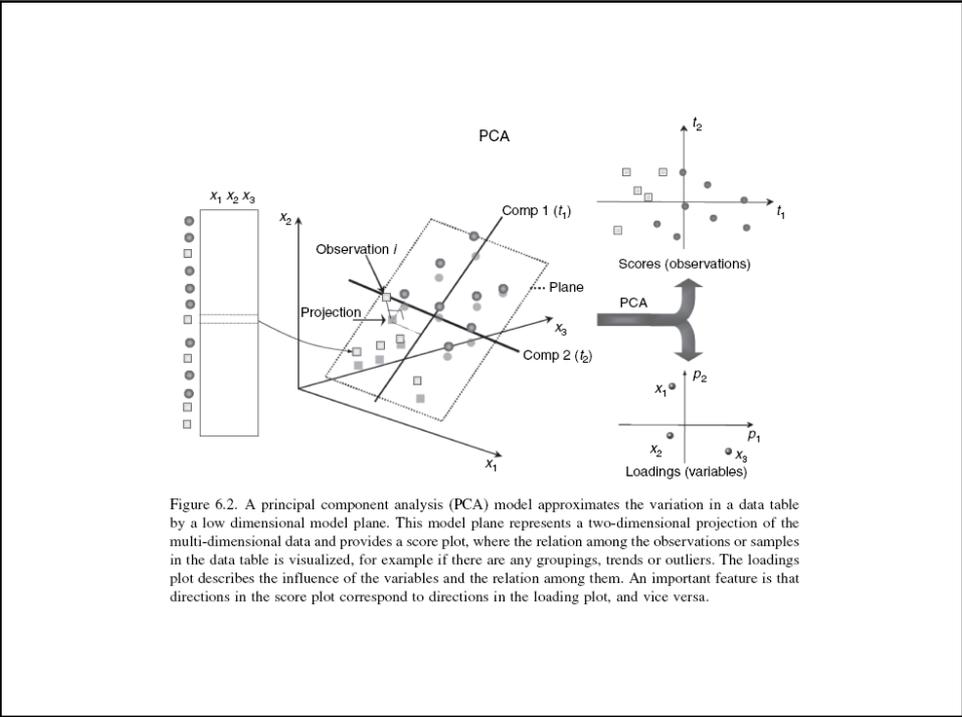
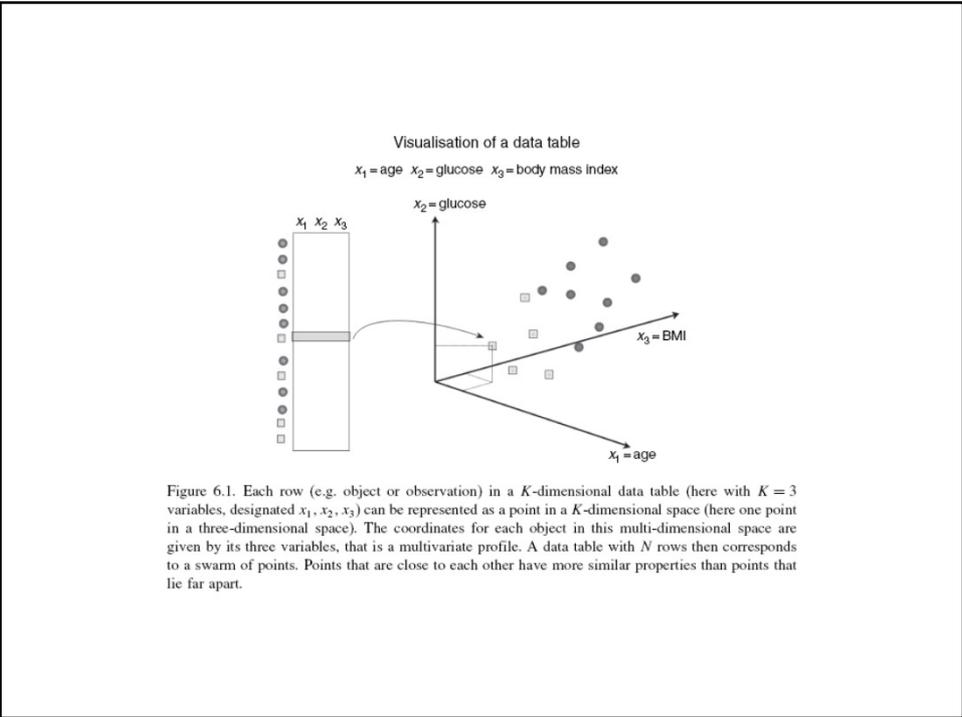


Table 5.2 Case study 2

Element	Group	Melting point (K)	Boiling point (K)	Density (mg/cm ³)	Oxidation number	Electronegativity
Li	1	453.69	1615	534	1	0.98
Na	1	371	1156	970	1	0.93
K	1	336.5	1032	860	1	0.82
Rb	1	312.5	961	1530	1	0.82
Cs	1	301.6	944	1870	1	0.79
Be	2	1550	3243	1800	2	1.57
Mg	2	924	1380	1741	2	1.31
Ca	2	1120	1760	1540	2	1
Sr	2	1042	1657	2600	2	0.95
F	3	53.5	85	1.7	-1	3.98
Cl	3	172.1	238.5	3.2	-1	3.16
Br	3	265.9	331.9	3100	-1	2.96
I	3	386.6	457.4	4940	-1	2.66
He	4	0.9	4.2	0.2	0	0
Ne	4	24.5	27.2	0.8	0	0
Ar	4	83.7	87.4	1.7	0	0
Kr	4	116.5	120.8	3.5	0	0
Xe	4	161.2	166	5.5	0	0
Zn	5	692.6	1180	7140	2	1.6
Co	5	1765	3170	8900	3	1.8
Cu	5	1356	2868	8930	2	1.9
Fe	5	1808	3300	7870	2	1.8
Mn	5	1517	2370	7440	2	1.5
Ni	5	1726	3005	8900	2	1.8
Bi	6	544.4	1837	9780	3	2.02
Pb	6	600.61	2022	11340	2	1.8
Tl	6	577	1746	11850	3	1.62

PCA

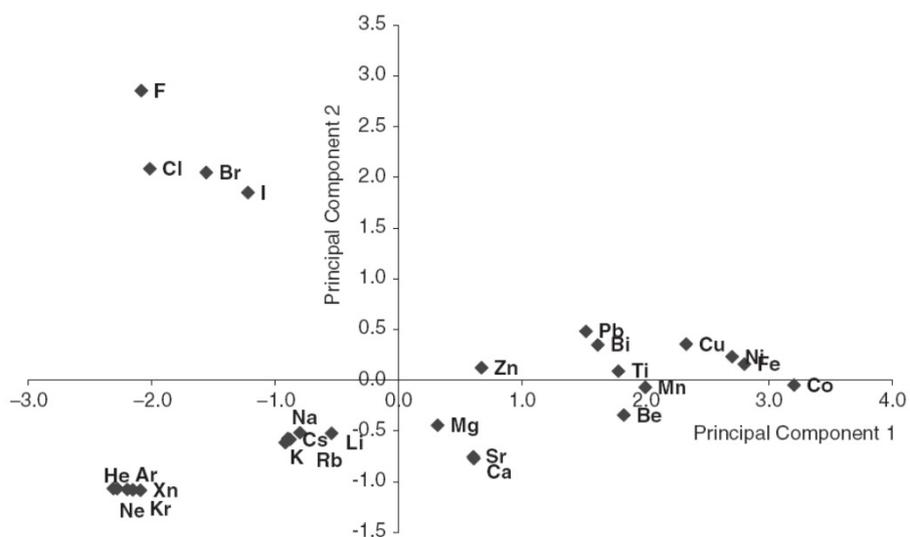
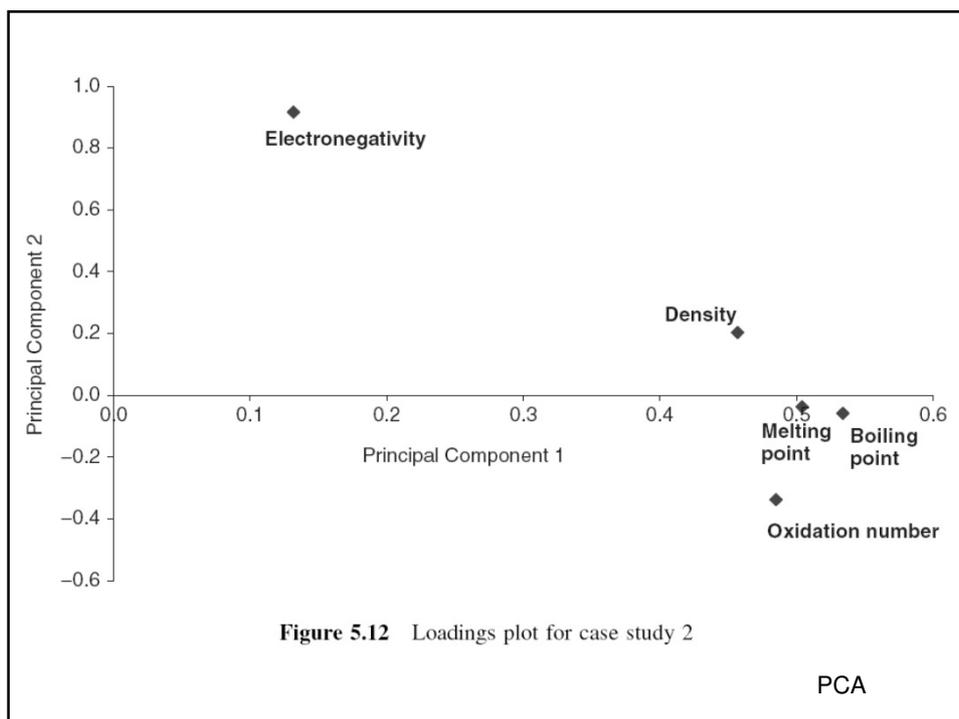


Figure 5.9 Scores plot of the first two PCs for case study 2

PCA



Cluster analysis

Table 5.4 Simple example for cluster analysis; six objects (1–6) and seven variables (A–G)

Objects	Variables						
	A	B	C	D	E	F	G
1	0.9	0.5	0.2	1.6	1.5	0.4	1.5
2	0.3	0.2	0.6	0.7	0.1	0.9	0.3
3	0.7	0.2	0.1	0.9	0.1	0.7	0.3
4	0.5	0.4	1.1	1.3	0.2	1.8	0.6
5	1.0	0.7	2.0	2.2	0.4	3.7	1.1
6	0.3	0.1	0.3	0.5	0.1	0.4	0.2

Table 5.5 Correlation matrix for the six objects in Table 5.4

	1	2	3	4	5	6
1	1					
2	-0.338	1				
3	0.206	0.587	1			
4	-0.340	0.996	0.564	1		
5	-0.387	0.979	0.542	0.990	1	
6	-0.003	0.867	0.829	0.832	0.779	1

CA

Table 5.6 First step of clustering of data from Table 5.5, with the new correlation coefficients indicated as shaded cells, using nearest neighbour linkage

	1	2 and 4	3	5	6
1	1				
2 and 4	-0.338	1			
3	0.206	0.587	1		
5	-0.387	0.990	0.542	1	
6	-0.003	0.867	0.829	0.779	1

CA

